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For example, look at answer #8. Note that the 58.443 is in the denominator on the right side and you generate the final answer by doing 0.200 times 0.100 times 58.443. Note that the 58.443 is in the denominator on the right side and you generate the final answer by doing 0.200 times 0.100 times 58.443.

### **ChemTeam: Molarity Problems #1 - 10**

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Page 1 of 2 1. Sea water contains roughly 28.0 g of NaCl per liter.

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Molarity Worksheet W 331 Everett Community College Student Support Services Program What is the molarity of the following solutions given that: 1) 1.0 moles of potassium fluoride is dissolved to make 0.10 L of solution. 2) 1.0 grams of potassium fluoride is dissolved to make 0.10 L of solution. 3) 1.0 grams of potassium fluoride is dissolved to make 0.10 mL of solution. 4) 952 grams of ...

### **Molarity Worksheet W 331 - Everett Community College**

Inorganic Nomenclature Worksheet 1. ammonium sulfide 51. aluminum acetate 101. sodium acetate 2. sodium nitrate 52. calcium chloride dihydrate 102. zinc sulfite

### **Inorganic Nomenclature Worksheet - ChemTeam**

6. Calculate the volume of a 1.25 M solution of HCN made from 31.0 grams of HCN. (Answer: 0.919 Liters).  
7. Calculate the volume of a 3.50 molar solution of H<sub>2</sub>SO<sub>4</sub>

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